



SE-6887

**B. E. III (Sem. V) (Textile Processing) Examination**  
**April / May – 2011**  
**Chemistry of Surface Active Agents**

Time : 3 Hours]

[Total Marks : 100

**Instructions :**

(1)

नीचे दृष्टावेक निशानीवाणी विगतो उत्तरवडी पर अवश्य बपवी. Fillup strictly the details of signs on your answer book.	Seat No. :
Name of the Examination :	<input type="text"/>
<input type="checkbox"/> B. E. 3 (Sem. 5) (Tex. Proc.)	<input type="text"/>
Name of the Subject :	<input type="text"/>
<input type="checkbox"/> Chemistry of Surface Active Agents	<input type="text"/>
Subject Code No. : <input type="text" value="6"/> <input type="text" value="8"/> <input type="text" value="8"/> <input type="text" value="7"/>	<input type="text"/>
Section No. (1, 2,.....) : <input type="text" value="1"/> <input type="text" value="2"/>	<input type="text"/>
	Student's Signature

- (2) Answers to the two sections must be written in same answer book.  
(3) Figures to the right indicate full marks.

**SECTION - I**

- 1 (a) Answer the following objective questions : 10
- (1) Carboxylate soaps get precipitated by addition of alkali. - True/False, correct if false.
  - (2) Why mercury does not wet the glass ?
  - (3) Anionic softeners are not widely used for cotton fabric finishing - True/False, Justify.
  - (4) SLS is a nonionic surfactant, true or false ? Correct if false.
  - (5) What are micelles ?
  - (6) What is difference between levelling agent and retarding agent ?
  - (7) State the mathematical expression for surface tension.
  - (8) Define : H.L.B.
  - (9) What is meant by NP-20 ?
  - (10) Give the chemical name and structure of Dedanol OT.

- (b) Explain what is meant by CMC and state the effect of chemical structure on CMC. **10**
- 2** (a) Classify anionic surfactants. Discuss the chemistry, properties and applications of carboxylates. **10**
- (b) Explain with proper illustrations the application of surfactant as rubbing fastness improver. **5**

**OR**

- 2** Write a critical review on the theory of surface activity and hydrophilic-lyophilic balance. **15**
- 3** Write short notes : (any **three**) **15**
- (a) Speciality surfactants
- (b) Finishing softeners
- (c) Phosphate esters
- (d) Emulsification.

**SECTION - II**

- 4** (a) Answer the following objective questions : **10**
- (1) \_\_\_\_\_ is the quantitative expression for surface tension.
- (2) Sodium stearate is a carboxylate soap, true or false ?
- (3) Which surfactant is available as a byproduct of paper industry ?
- (4) Give the reaction scheme for the action of fixanol C in direct dyeing of cotton.
- (5) What is the ionic nature of dye fixing agents used for direct dyes ?
- (6) What is dedenol OT paste chemically ?
- (7) Why ethoxylates possess cloud point ?
- (8) Cloud point of ethoxylates \_\_\_\_\_ with increase in EO content.
- (9) Name one non-surfactant based levelling agent used in acid dyeing of nylon.
- (10) N.P. 20 can act as a good solubelizer, true or false ?
- (b) Describe various technical aspects of foaming. **10**

- 5** Give characteristic properties of non-ionic surfactants. **15**  
Discuss the chemistry, properties and applications of nonyl phenol ethoxylates.

**OR**

- 5** Give a general classification of surfactants. Discuss **15**  
chemistry, properties and applications of cationic surfactants.
- 6** Write short notes : (any **three**) **15**
- (a) Dye fixing agent
  - (b) Amphoteric surfactants
  - (c) Solubility and Kraft point
  - (d) Sulphosuccinates.

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